# EFFECT OF ISOTHERMA L AGING ON THE INTERFACIAL REA CTIONS BETWEEN SN1.5AG0.7CU9.5IN SOLDER AND CU SUBSTRATE

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Abstract: The aim of article is study theeffect of isothermal aging on theinterfacial reactions between \$1.5Ag0.709.5In solder and \$\mathbb{Q}\$ is substrate. Aging of joints made with tinbased alloy \$1.5Ag0.7Cu9.5In was carried out at various temperatures ranging to \$10\circ\$C, at various times ranging to \$10\circ\$C, at vari

Key words:Lead-freesolder Intermetallic compound,Aging, Interfacial reaction

### 1. INTRODUCTION

The rapid development of electronic packaging and assembly termology has resisted in a demand for advancement in the performance of soletring interconnects (Vi áš, et al., 2009). Reliability of soldering interconnects in eletronic packaging ad assembly is primarily determined by the microstructure of interconnects, whereashe microstructure of interconnects is dependent in the interfaial reaction during soldering and operation solvice (Zeng te al. 2010). In this article, the microstructure, especially theinterfacial reaction behaviors betveen leal-fiee solder Sn-Ag-Cu-In and the Cu substrate inaging processwas studied and mechanisms for certain reactin phenomenonwere investigated. Currently from lead-free solers the attention is getting the cutectic SnAcallov. It has a highemelting pointthan SnPb addy and no risk to the environment. Systems of leathfree solders are mainly based on the addition of a small amount of third or fourth addition in binary alloysto improve thei properties. Idium is addedo the solder alloys in order to decease melting pint and improve the wettability and this make possible of lower soldering temperature Kanlayasiri, 2009). The addion of indium also advances the xidation resistance (Šeboteal., 2009; Vi áš et al., 2008).

## 2. EXPERIMENT

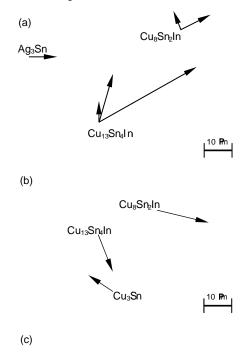
The four component læc-free solderSn1.5Ag0.7Q9.5In was created b casting, gradally melted from pure metal and soldering allows by induction heating in  $A_2O_3$  crucible. Metals and alloys were used in theorm of ingot Sn 99.99% ad wire (SnAg5.0, Scu3.0). Individual components were weighed by 0.01g accurace Technical opper of purity 99.99 % was used as the basic material. Soldeing was provided by using the hot plate methodat temperature of 250 Cfor 4s. The ceated samples of solder joints Cu-Sn1.4Sn.07Cu9.5In were subsequently aged at temperatures of 130,150 and 170°C for 16 days. The samples were gradually taken away from the vacuum furace at interval of 2, 4, 8, 12 and 16 days. For microstructural analysis theheat-affected rad unaffected joints samples were grinded and polished with diamond pasteup to particle sizeof 0.7 m and then were effect (2% HCl+5%).

HNO<sub>3</sub>, 93% methaol) for 2-4 s.The microstruture of soldered joints and morphlogy of internetallic phase presented ni solder structure and on the joints interface was investigated to optical microscopy. To evaluate the chemical composition and identify different phases the Ex microanalysis (JEOL-JXA 840A) was carried but.

#### 3. RESULTS AND DISCUSSION

3.1 Evolution of IMFs at the interface between the solde bulk and Cu substrate during aging

The microstructure of Cu-Sn1.5Ag0.7Cu9.5n solder joins interface after soldering and hat affecting attemperature to 150°C is shown infigure 1.



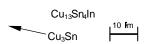


Fig. 1. Microstructural evolution of Cu-Sn15Ag0.7Cu9.5h solder joints aged t 150°C for valious times; () as-soldred, (b 48 h, (c) 384 h.

The structure of the solder Sn1.5Ag07Cu9.5In after soldering is characterized by beterogeneity. The volume 6 solder are three passes, the firstAg<sub>3</sub>Sn consist 6Ag element in

the used solder composition, which during influence of heat and time changes its shape and size. The second phase Cu<sub>8</sub>Sn<sub>2</sub>In which is dispersed in the Sn matrix of solder, and phase Cu<sub>13</sub>Sn<sub>4</sub>In, which is located near the interface. Due to the high solubility of In in the Sn at the Cu-Sn1.5Ag0.7Cu9.5In interface after soldering was observed the ternary Cu<sub>13</sub>Sn<sub>4</sub>In intermetallic phase. The presence of this phase was confirmed by EDX microanalysis (58.46 at.% Cu, 33.12 at.% Sn, at 8.41% In). The thickness of IMC layer was about 2 μm. There can be seen the next intermetallic phase Cu<sub>3</sub>Sn after aging at Cu -Cu<sub>13</sub>Sn<sub>4</sub>In interface (Fig. 1b). Morphologes of the IMCs at the interface are significantly different each other. IMC Cu<sub>13</sub>Sn<sub>4</sub>In is initially consisted of several sprouts (indent, spicules) differently oriented into the solder, incomparises to Cu<sub>3</sub>Sn phase. The indented shape of Cu<sub>13</sub>Sn<sub>4</sub>In phase had changed over time to highly asymmetric one with plenty thick sprouts, being seen especially at the aging temperatures of 150 and 170°C, when the phase grows relatively quickly (Fig. 1c).

The EDX microanalysis of heat affected structure of solder joints Cu-Sn1.5Ag0.7Cu9.5In with created phases can be observed in figure 2. The line analysis confirmed the presence of reaction products at the interface and the presence of phases Ag<sub>3</sub>Sn and Cu<sub>8</sub>Sn<sub>2</sub>In in the solder and on the interface.

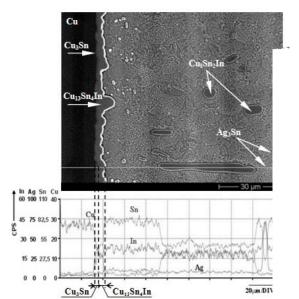


Fig. 2. EDX line profile of Cu–Sn1.5Ag0.7Cu9.5In solder joint interface aged at  $150^{\circ}$ C for 28 h

# 3.2 Growth rate of interfacial IMCs during aging

The direction of intermetallic phase's growth, their thickness as dependence on the aging time at different temperatures is documented in figure 3. The results show that temperature increase causes increase of IMC. The most significant increase of layer thickness can be observed at the  $Cu_{13}Sn_4In$  phase which is growing faster at all temperatures than  $Cu_3Sn$ . At temperatures 130 and 150°C the  $Cu_3Sn$  intermetallic phase has the linear kinetics of growth with slower growth compared at the temperature 170°C. Intermetallic phase thickness growth as time dependence seems to be linear at the beginning, but after eight days at 170°C this linear character disappears.

## 4. CONCLUSION

In this article, the microstructure, especially the interfacial reaction behavior between lead-free solder Sn1.5Ag0.7Cu9.5In and the Cu substrate in aging process was studied. Under the impact of the thermal action it is possible to observe two intermetallic layers darker Cu<sub>3</sub>Sn and lighter Cu<sub>13</sub>Sn<sub>4</sub>In. Noncontinuous grouping of Cu<sub>8</sub>Sn<sub>2</sub>In IMC can be observed, too.

Fig. 3. Graph of the average thickness of the intermetallic layer Cu<sub>3</sub>Sn and Cu<sub>13</sub>Sn<sub>4</sub>In at the Cu/Sn1.5Ag0.7Cu9.5In interface in dependence of aging time.

The thickness of the  $Cu_{13}Sn_4In$  layer shows almost linear dependence of the time at beginning. The growth rate of the  $Cu_3Sn$  layer reveals the character corresponding to the diffusion process. The average thickness of both IMC layers reached the value of 42.6  $\mu$ m at 150°C. Intermetallic compounds had physical and mechanical properties different from the solder bulk and substrates, so an excess of IMC would be of considerably affect the fatigue strength and fracture strength of the interconnects. Future research activities assessing the influence of thickness IMC on thermomechanical fatigue of solder joints.

# 5. ACKNOWLEDGEMENTS

This paper was supported under projects VEGA 1/0381/08 and VEGA 1/0111/10.

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